

ALKALOIDS OF THE PAPAVERACEAE. VII.¹ THE ABSOLUTE CONFIGURATION
OF ROMNEINE. (-)-RETICULINE FROM ROMNEYA COULTERI.

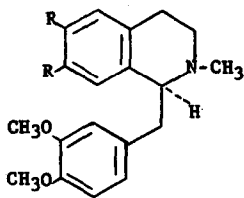
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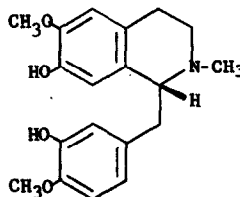
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The isolation and structure proof by synthesis of romneine from Romneya coulteri var. trichocalyx was recently described³. The natural alkaloid, isolated as an oil, had $[\alpha]_D^{27} = +37^\circ$ (c 0.11, ethanol)⁴. Attempts to resolve synthetic romneine in order to obtain material for chirality determination failed. By use of the standard isolation procedure³ with additional large amounts of plant material, romneine was isolated as the hydrobromide salt (m.p. 226-227°; $[\alpha]_D^{28} = +40^\circ$, c 0.26, ethanol). The alkaloid salt was treated with sulfuric acid and phloroglucinol⁵ and the resultant product (after being converted to the free base) was methylated with diazomethane to yield laudanosine (Ia), $[\alpha]_D^{28} = +83^\circ$ (c 0.091, ethanol), (lit.⁶ $[\alpha]_D^{22} = +105^\circ$, ethanol). Since (+)-laudanosine has been shown⁷ to be Ia, (+)-romneine is established as Ib.⁸

Work-up of the combined pH 9 fractions³ from extraction of several kilograms of R. coulteri yielded, after chromatography on basic alumina, 50 mg of an amorphous white powder, m.p. 78-90°, which was homogeneous by thin-layer chromatography and which had identical UV, IR, and NMR spectra and R_f value as an authentic sample of (+)-reticuline. However, the optical rotation was of the opposite sign: $[\alpha]_D^{28} = -55^\circ$ (c 0.44, ethanol)⁸. This represents the first isolation of reticuline in which the (-)-isomer (II) predominates, although other



Ia, R = OCH₃
b, R - R = -OCH₂O-



II

papaveraceous species have been found to contain an excess of (+)-reticuline along with (+)-reticuline. It is presumed that the low optical rotation value of our isolated reticuline (lit.⁹ value for pure (+)-reticuline: $[\alpha]_D = +132^\circ$, methanol) indicates the presence of (+)-reticuline since it was above shown that R. coulteri contains romaine which is configurationally related to (+)-reticuline. We did not have sufficient material so that an actual separation of the isolated reticuline into the pure (-)- and (+)-forms could be carried out.

REFERENCES

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3. F. R. Stermitz, L. Chen, and J. White, Tetrahedron 22, 1095 (1966).
4. This is a correction to the value originally reported³.
5. H. R. Arthur, W. H. Hui, and Y. L. Ng, J. Chem. Soc. 1959, 1840.
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8. We are indebted to Professor John C. Craig (University of California, San Francisco) for confirming, by optical rotatory dispersion, the absolute configurations assigned here.
9. F. Kusada, Pharm. Bull. (Japan) 1, 189 (1953).